

# ZACLON LLC MATERIAL SAFETY DATA SHEET

## IDENTIFICATION

**NAME:** Zaclon\* High Temperature Galvanizing Flux Solution

**GRADE:** W-HT

**CHEMICAL FAMILY:** Inorganic Salt Solution

**MANUFACTURER /DISTRIBUTOR:**  
Zaclon LLC

**PRODUCT INFORMATION PHONE:**  
(800) 356-7327

**ADDRESS:**  
2981 Independence Road  
Cleveland, Ohio 44115

**TRANSPORTATION EMERGENCY PHONE:**  
CHEMTREC (800) 424-9300

## PHYSICAL DATA

**BOILING POINT, 760 mm Hg**  
>100 °C (>212 °F)

**MELTING POINT**  
Not determined.

**SPECIFIC GRAVITY**  
1.35 TO 1.45

**VAPOR PRESSURE**  
24 mm Hg at 20 °C (68 °F)  
9 mm HG at 37.7 °C (100 °F)

**VAPOR DENSITY**  
Vapor is water.

**SOLUBILITY IN WATER**  
100%

**pH INFORMATION**  
2.5 TO 3.0

**EVAPORATION RATE**  
BUTYL ACETATE=1) <1

**FORM**  
Liquid

**APPEARANCE**  
Clear

**COLOR**  
Colorless to water white

**ODOR**  
Odorless

\*Reg. U.S. Pat. & Tm. Off., Zaclon LLC Zaclon® Galvanizing Fluxes are made only by Zaclon LLC

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The data in the Material Safety Data Sheet relates only to the specific material designated herein and does not relate to use in combination with any other material or in any process.

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**HAZARDOUS COMPONENTS**

<b><u>MATERIAL(s)</u></b>	<b><u>CAS #</u></b>	<b><u>APPROXIMATE %</u></b>
Zinc Chloride	7646-85-7*	40
Ammonium Chloride	12125-02-9**	20
Magnesium Chloride	7786-30-3	1 - 10

\* Regulated as a Toxic Chemical under Section 313 of Title III/SARA, and 40 CFR Part 372.

\*\*Ammonium Salts Generate Ammonia in Water. For SARA 313 Requirements, 10% of the Total ammonia present in aqueous ammonia solutions is reportable.

**NON-HAZARDOUS COMPONENTS**

<b><u>MATERIAL(s)</u></b>	<b><u>CAS #</u></b>	<b><u>APPROXIMATE %</u></b>
Water	7732-18-5	45

**HAZARDOUS REACTIVITY**

**INSTABILITY** Stable

**INCOMPATIBILITY** Incompatible with cyanide and sulfide salts (may release toxic gases).

**DECOMPOSITION** At high temperatures, as in intended use, ammonium chloride fumes, zinc oxide fumes, zinc chloride fumes, and ammonia and hydrogen chloride gases may be released.

**POLYMERIZATION** Will not occur.

**FIRE AND EXPLOSION DATA**

**FLASH POINT** Will not burn.

**FLAMMABLE LIMITS IN AIR, % BY VOL.**

**LOWER** Not applicable. **UPPER** Not applicable

**AUTOIGNITION TEMPERATURE** Not Applicable.

**AUTODECOMPOSITION TEMPERATURE** Not Available.

**FIRE AND EXPLOSION HAZARDS**

After drying, may release zinc oxide, zinc chloride, and ammonium chloride fumes and ammonia and hydrogen chloride gases at high temperatures.

**EXTINGUISHING MEDIA** As appropriate for combustibles in area.

**SPECIAL FIRE FIGHTING INSTRUCTIONS** None.

## **HEALTH HAZARD INFORMATION**

**PRINCIPAL HEALTH HAZARDS** (Including Significant Routes, Effects, Symptoms of Over exposure, and Medical Conditions Aggravated by Exposure)

Causes eye burns. Causes skin irritation or burns.  
Causes irritation of lungs and upper respiratory passages.  
Inhalation 10 minute LC50 for zinc chloride: 2000 MG/M<sup>3</sup> in rats.  
Oral LD50 for zinc chloride: 1100 mg/kg in rats.  
Oral LD50 for ammonium chloride: 1650 mg/kg in rats.

The product is corrosive to the eyes and corrosive or irritating to skin. Toxic effects described in animals from short exposures include corrosion of mucosal surfaces, liver effects, and kidney effects. Toxic effects in animals occurring only with inhalation exposures are lower respiratory infection with pulmonary edema. Tests in bacterial or mammalian cell cultures demonstrate mutagenic activity. Tests in some animals indicate that the compound may have embryotoxic activity.

Human health effects of overexposure by inhalation, ingestion, or skin or eye contact may initially include: eye irritation with discomfort, tearing, or blurring of vision; skin irritation with discomfort or rash; or irritation of the upper respiratory passages. Higher exposures may lead to these effects: skin and eye burns or ulceration; temporary lung irritation effects with cough, discomfort, difficulty breathing, or shortness of breath; possibly modest initial symptoms, followed in hours by severe shortness of breath, requiring prompt medical attention; fatality from gross overexposure by fume inhalation or by significant ingestion. There are inconclusive or unverified reports of human sensitization. Individuals with preexisting diseases of the lungs may have increased susceptibility to the toxicity of excessive exposures.

When the Zaclon® products are heated to high temperatures as those encountered in the galvanizing process, irritating zinc chloride fumes and gaseous hydrogen chloride may be released. Severe exposures may cause pulmonary edema. Heating may also release zinc oxide fumes which may cause metal fume fever.

### **CARCINOGENICITY**

No ingredient listed as a carcinogen by IARC, NTP, OSHA, or ACGIH.

### **EXPOSURE LIMITS (PEL (OSHA), TLV (ACGIH), ETC.)**

Zaclon® fluxes are not listed as hazardous materials by OSHA or ACGIH. After drying, however, zinc chloride, zinc oxide, ammonium chloride fumes and hydrogen chloride gas may be released at high temperatures. The OSHA 8-hour Time Weighted Average (TWA) and ACGIH TLV®-TWA limits for these products are: zinc chloride = 1 mg/m<sup>3</sup>; zinc oxide = 5 mg/m<sup>3</sup>; hydrogen chloride = 5 ppm or 7 mg/m<sup>3</sup> (ceiling). Exposure limits for ammonium chloride have not been established by OSHA. The ACGIH 8 hour TLV®-TWA is 10 mg/m<sup>3</sup> for ammonium chloride fumes and the ACGIH Short Term Exposure Limit (STEL) is 20 mg/m<sup>3</sup> for ammonium chloride fumes.

### **SAFETY PRECAUTIONS**

Do not get in eyes, on skin, on clothing.  
Avoid breathing mists or fumes.  
Wash thoroughly after handling.

**FIRST AID**

In case of contact, immediately flush eyes or skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Call a physician. Wash contaminated clothing before reuse and discard shoes.

If inhaled, remove to fresh air immediately. If not breathing, give artificial respiration, preferably mouth-to-mouth. If breathing is difficult, give oxygen. Call a physician.

If swallowed, do not induce vomiting. Give large quantities of water. Call a physician immediately. Never give anything by mouth to an unconscious person.

**NOTE TO PHYSICIAN**

Emesis should be initiated unless there is evidence of burns of the oral mucosa. Gastric lavage may be advisable. In some cases of excess body loads of zinc, treatment with calcium disodium acetate may be effective.

**PROTECTION INFORMATION****GENERALLY APPLICABLE CONTROL MEASURES**

Good general ventilation should be provided to keep fume concentrations below the exposure limits.

**PERSONAL PROTECTIVE EQUIPMENT**

Coverall chemical splash goggles and rubber gloves, and apron should be worn for routine work. A full-length face shield should be worn around galvanizing kettles. If considerable contact is likely, wear impervious (rubber) clothing or acid suit. If air concentrations exceed exposure limits, use OSHA-permissible respiratory equipment.

**DISPOSAL INFORMATION****SPILL, LEAK OR RELEASE**

Comply with Federal, State and local regulations on reporting releases. Flush entire spill area with plenty of water to chemical sewer.

**WASTE DISPOSAL**

Comply with Federal, State, and local regulations. If approved, may be flushed to chemical sewer to transferred to a licensed disposal contractor.

### **SHIPPING INFORMATION**

#### **DOT/IMO**

**PROPER SHIPPING NAME:** Corrosive Liquid, Acidic, Inorganic, N.O.S.

(Contains Zinc Chloride and Zinc Ammonium Chloride)

**HAZARD CLASS** 8, Corrosive Material

**I.D. NUMBER** UN NO. 3264

**PACKING GROUP:** PG III

**REPORTABLE QUANTITY:** 1,000 lbs. (Zinc Chloride and Zinc Ammonium Chloride)

### **OTHER INFORMATION**

**SHIPPING CONTAINERS** Tank Cars, Tank Trucks, 55 Gallon Drums, Bulk Drums (Totes).

**STORAGE CONDITIONS** Keep containers tightly closed. Do not store with cyanides or sulfides.

#### **NPCA - HMIS RATINGS**

Health 3

Flammability 0

Reactivity 0

Personal Protection \*

\*Personal protection rating to be supplied by user depending on use conditions.

#### **TITLE III HAZARD CLASSIFICATION:**

Acute Yes

Chronic No

Fire No

Reactivity No

Pressure No

### **ADDITIONAL INFORMATION AND REFERENCES**

For further information see Zaclon's Zaclon® Galvanizing Flux Data Sheet.

**DATE OF LATEST REVISION/REVIEW:** May 2008

**PERSON RESPONSIBLE FOR MSDS:** J.L. Curry

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